2040 - The Maths of Carbon



| Name | Class | |
|------|-------|--|
| | | |

Student Worksheet

Thought-starters: What's the deal with Carbon Dioxide? Surely an invisible, tasteless and smell-less gas can't really be such a drama? Can it?

The Issue of Atmospheric Carbon

Watch this clip:



2040 - Exploring the Themes Password: 2040_EDU

While you're watching, complete the table below.

| SEE – What did you see as you watched this video? | HEAR – What did you hear the narrator talk about in the video? | WONDER – What questions arose as you watched? |
|---|--|---|
| | | |

After watching, complete Column A. Then wait for further instructions.

- Column A There were many ideas presented in this clip, by the narrator, about the problem we face today regarding the Earth's atmosphere.
 What THREE ideas did you find most interesting?
- Column B Share your thoughts in column A with a partner and note down anything new.
- Column C What were some ideas shared in the class discussion that you hadn't considered before?

| Column A | Column B | Column C |
|----------|----------|----------|
| | 1. | |
| | | |
| | | |
| | | |
| | 2. | |
| | | |
| | | |
| | 3. | |
| | | |
| | | |
| | | |

The Air That I Breathe...

In the table below, list each of the main gases in our atmosphere as mentioned by your teacher.

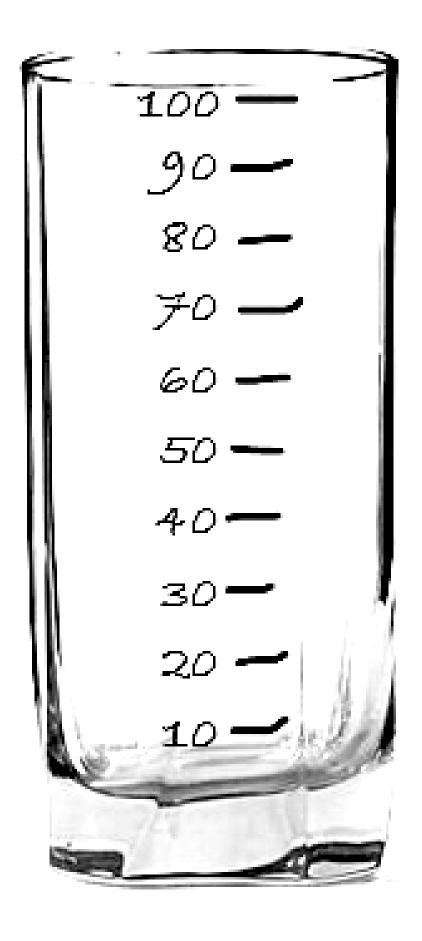
After chatting with your class, then list the amount of that gas as a proportion of the total atmosphere - first, as 'parts per million' (PPM) and then as a percentage:

(Remember, to calculate a percentage (%) from 'Parts per Million' (PPM): $\frac{PPM/1\ 000\ 000x}{100 = _____$ %)

| Name of Gas | Amount in Atmosphere - as 'Parts per Million' (PPM) | Amount in Atmosphere - as a Percentage (%) |
|-------------|--|--|
| | | |
| | | % |
| | | |
| | | |
| | | % |
| | | |
| | | % |
| | | 70 |
| | | |
| | | % |
| | | |
| | | |

| | % |
|--|---|
| | |

Optional - If the Earth's Atmosphere was a glass of frozen slushie drink, and each gas was a separate coloured 'flavour' draw and label the components in the glass! (Remember, in the real atmosphere, all the gases are mixed in together - not in layers!)



Reflection

After completing the graphing activity, complete this activity.

| THINK – What are some thoughts you have about the activities and data in this lesson? | FEEL – How do you feel about what you have learned in this lesson? | WONDER – After this lesson, what are you still wondering about? |
|---|--|---|
| | | |
| Write three important ideas that 1. | you will take away from this le | sson: |
| 2. | | |
| | | |

| , | 3. | |
|---|----|---|
| | | |
| _ | | _ |
| | | |
| _ | | _ |
| | | |
| | | _ |
| | | |

These lessons have been created in partnership with

2040, Good Thing Productions



